

Review Exercises 1:

#1 Mass Reactants = Mass Products

$$11\text{g} = \underline{11\text{g}}$$

#2 Mass Reactants = Mass Products

$$8\text{g} + 64\text{g} = \underline{72\text{g}}$$

#3 Copperoxide + Carbon \rightarrow Carbon Dioxide + Copper

$$30\text{g} + 20\text{g} \rightarrow \mathbf{x} + 42\text{g}$$

$$50\text{g} \rightarrow 50\text{g}$$

$$\mathbf{x = 8g}$$

Review Exercises 2:

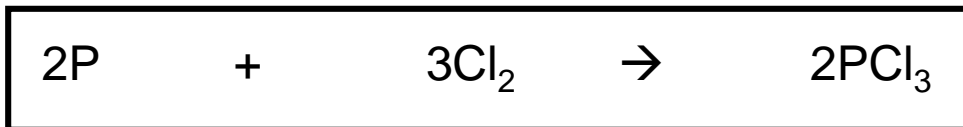
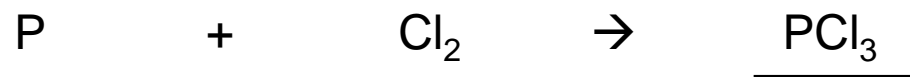
#1 d)

#2 d)

#3 c)

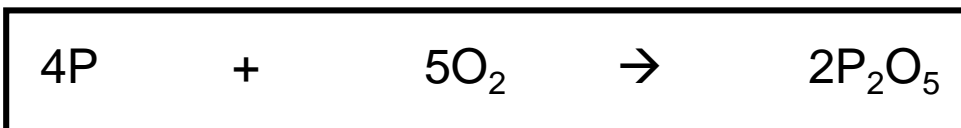
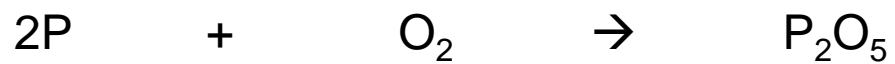
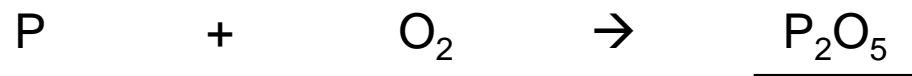
Review Exercises 3:

#1 a)



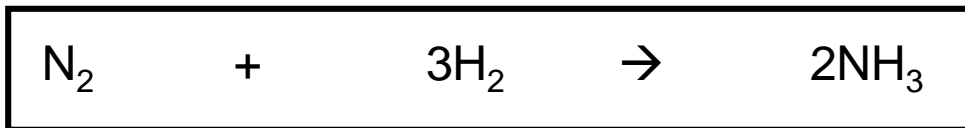
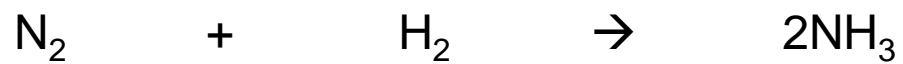
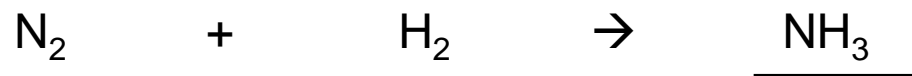
Review Exercises 3:

#1 b)



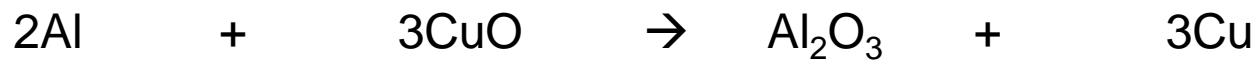
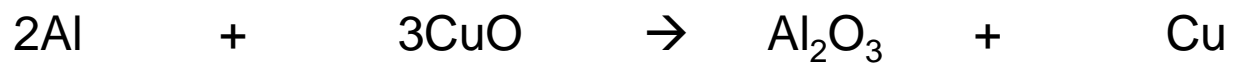
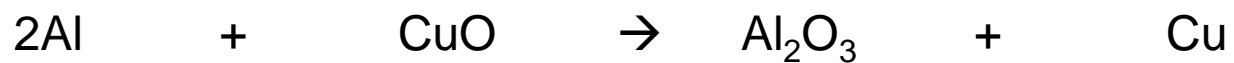
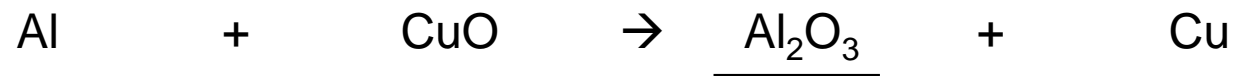
Review Exercises 3:

#1 c)



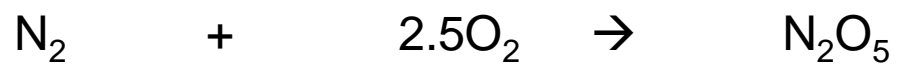
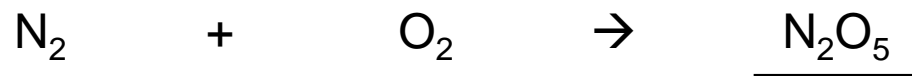
Review Exercises 3:

#1 d)



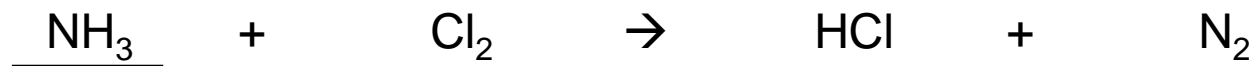
Review Exercises 3:

#1 e)



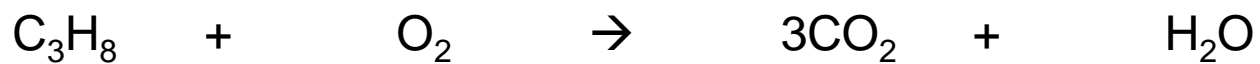
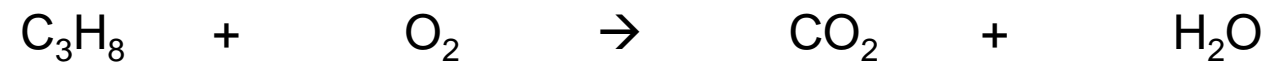
Review Exercises 3:

#1 f)



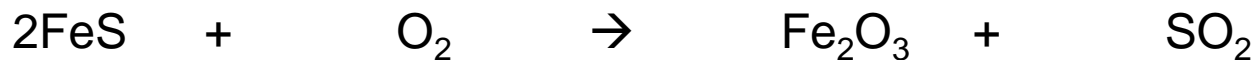
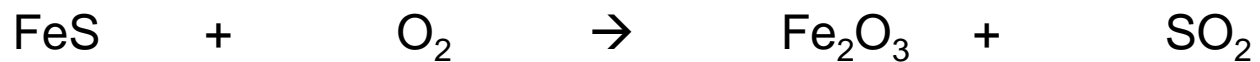
Review Exercises 3:

#1 g)



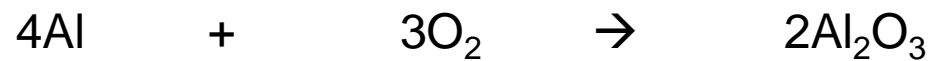
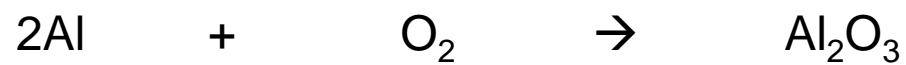
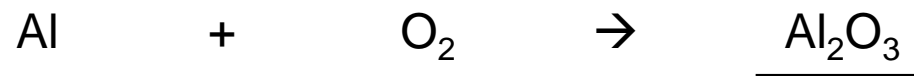
Review Exercises 3:

#1 h)



Review Exercises 3:

#2 a)



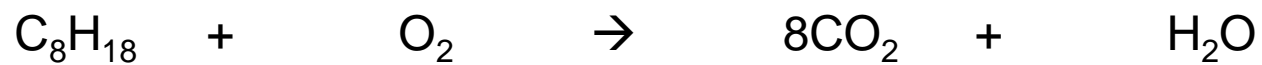
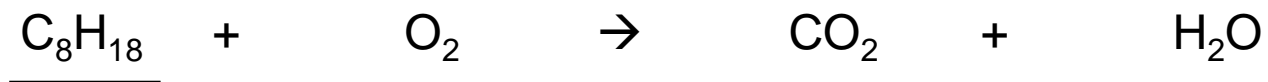
Review Exercises 3:

#2 b)



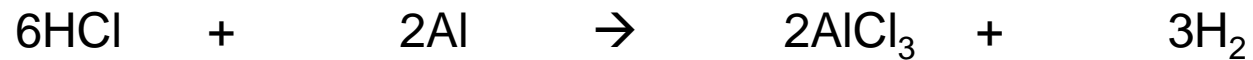
Review Exercises 3:

#2 c)



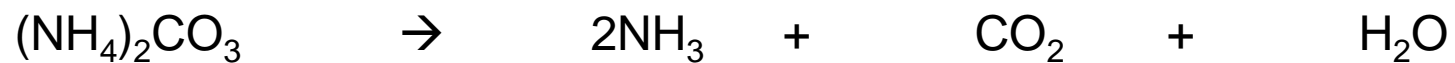
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#2 d)



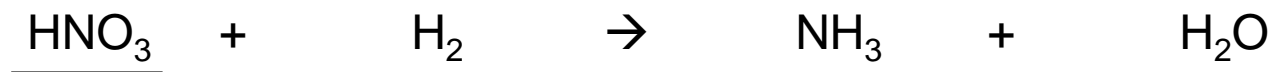
Review Exercises 3:

#2 e)



Review Exercises 3:

#2 f)



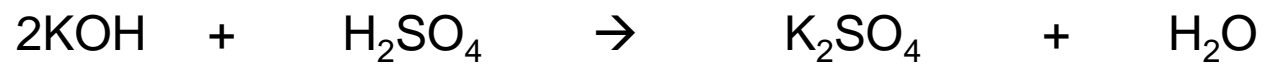
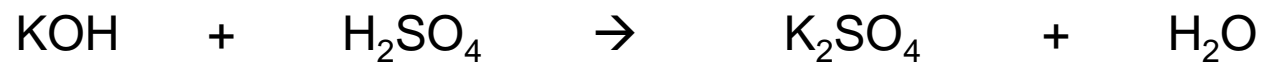
Review Exercises 3:

#2 g)



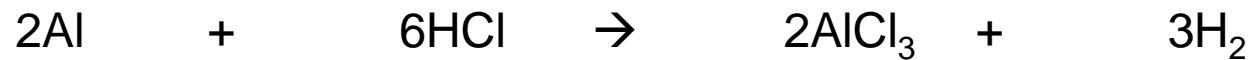
Review Exercises 2:

#3 a)



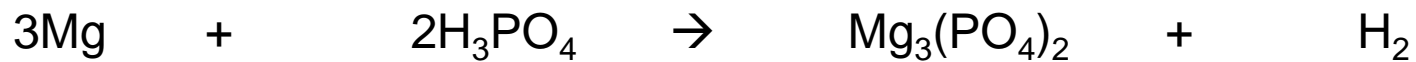
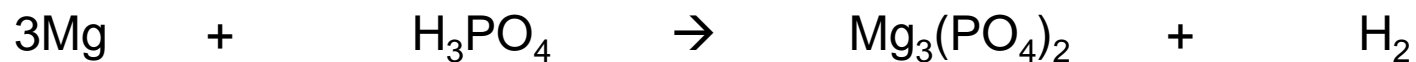
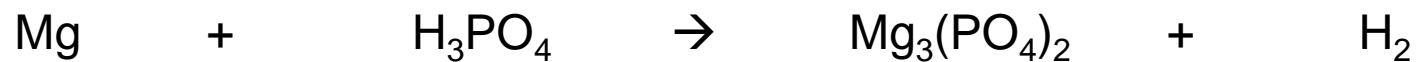
Review Exercises 2:

#3 c)



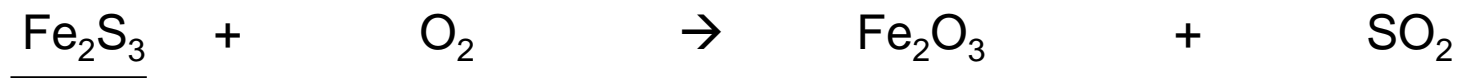
Review Exercises 2:

#3 d)



Review Exercises 2:

#3 e)



Review Exercises 2:

#3 f)



Review Exercises 2:

#3 g)

